

Fe-Ti-Y (Iron-Titanium-Yttrium)

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The two previous investigations of this ternary system by [1994Zen] and [1997Liu] reported isothermal sections at 500 and 600 °C respectively. These results were reviewed by [2001Rag] and [2000Rag]. Recently, [2009Gon] reported experimental results on this system at 1000 °C and presented a thermodynamic analysis.

Binary Systems

There are two intermediate phases in the Fe-Ti system: Fe_2Ti (C14, MgZn_2 -type hexagonal) and FeTi (B_2 , CsCl -type cubic). The Fe-Y system is characterized by the presence of four compounds: Fe_2Y ($C15$, MgCu_2 -type cubic), Fe_3Y (Be_3Nb -type rhombohedral), Fe_{23}Y_6 ($\text{Mn}_{23}\text{Th}_6$ -type cubic) and Fe_{17}Y_2 ($\text{Ni}_{17}\text{Th}_2$ -type hexagonal or $\text{Th}_2\text{Zn}_{17}$ -type rhombohedral). In the Ti-Y system, there

are no intermediate phases. The terminal solid solubility is negligible.

Ternary Phase Equilibria

The study of [2009Gon] presents serious discrepancies. Quoting from their paper: “The experimental results show that Fe_2Ti and Fe_2Y form a continuous solid solution, namely, $\text{Fe}_2(\text{Y},\text{Ti})$ ”. The previous results of [1994Zen] and [1997Liu] show little solubility between Fe_2Ti and Fe_2Y . Moreover, they form different crystal structures. [2009Gon] modeled the “continuous solid solution” phase with two sub-lattices, with Ti and Y sharing sites in the second sublattice. They computed two isothermal sections at 600 and 1000 °C. In none of the above sections, a continuous solid solution is seen. At 1000 °C, a large liquid field is indicated at the Ti corner. Ti is solid at this temperature and shows little solubility for Y. In view of these serious anomalies, the results of [2009Gon] are not discussed further.

The isothermal section at 500 °C from the first investigation of this system by [1994Zen] (not presented in the earlier reviews) is redrawn in Fig. 1.

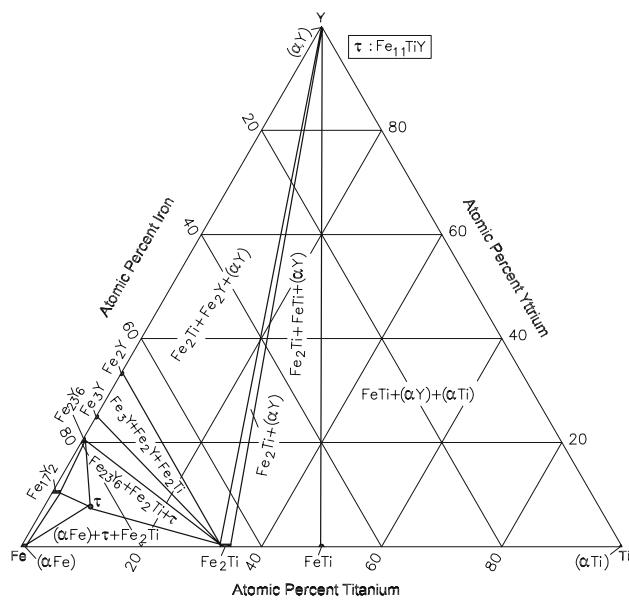


Fig. 1 Fe-Ti-Y isothermal section at 500 °C [1994Zen]

References

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